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OCCURRENCE OF ARSENIC, LEAD, THALLIUM AND BERYLLIUM IN GROUNDWATER

¹Abdul A.J. Mohamed, ¹Ibrahim Abdul Rahman, ²Sadri A. Said and ¹Lee H. Lim

¹Faculty of Science (FOS), Universiti Brunei Darussalam Jalan Tungku, Brunei Darussalam, BE1410, Brunei ²School of Pharmacy, College of Pharmacy and Nursing, University of Nizwa, Birkat Al Mouz, PB 33, PO 616, Nizwa, Sultanate of Oman

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ABSTRACT

The occurrence of carcinogenic and heavy metals in groundwater sources in Urban-west region of Zanzibar Island is an issue that is not very well known. This could be also coupled with the absence of drinking water treatment plants. This study for the first time reports on the occurrence and the levels of three carcinogenic metals-Arsenic (As), Beryllium (Be) and lead (Pb) in thirty groundwater samples collected from Zanzibar's Urban/West region. The levels of alkalinity, Magnesium (Mg) and Thallium (Tl) were also determined. The concentrations of As, Be, TI and Pb in the water samples were determined by the Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES). Palintest photometry procedures were used to determine the levels of total alkalinity and magnesium. Be, As, Tl and Pb were not detected (nd) in some water samples. The ranges of concentrations of Be, As, TI and Pb in the samples were; nd to 6100 ng L^{-1} , nd to 6600 ng L^{-1} , nd to 11600 ng L^{-1} and nd to 31400 ng L^{-1} respectively. The levels of total alkalinity varied from 38 to 380 (mg L⁻¹ as CaCO₃). The proportions of water samples contaminated with Be, Tl, As and Pb were 43.3, 66.7, 70 and 96.7% respectively. About 23% of the water samples had Pb concentrations beyond WHO limits for safe drinking water, while 30 and 56.67% of the samples had Be and Tl concentrations beyond the US EPA's maximum limits. The concentration of arsenic in each water sample was within WHO limits. The occurrence and the levels of carcinogenic metals in water sources could be a potential cause of cancer cases in Zanzibar. Therefore, prompt action is required to control the levels of these hazardous metals, and other possible contaminants in Zanzibar's domestic water systems.

Keywords: Carcinogenic Metals, WHO, Alkalinity, Prompt Action

1. INTRODUCTION

Access to clean and safe drinking water is a vital component in our social-economic life. Clean and safe water is a key factor for preventing the spread of waterborne and sanitation related diseases, such as cholera and dysentery; however large number of Zanzibar population still lack access to clean and safe water.

Most of the developing countries lack access to safe water and adequate sanitation infrastructures, which often results in the death of about two million infants annually (UNICEF, 2005; Cosgrove and Rijsberman, 2000; Gomez and Nakat, 2002).

Because of its polarity and ability to form hydrogen bonds, water acquires unique properties. It is also capable of absorbing, dissolving, adsorbing, or suspending an array of different compounds (WHO, 2007). The most common toxic elements in drinking water are As, Cd, Hg, Zn, Fe, Cl, Cr and Pb. The data and information for the water quality is vital in assessing its suitability for

Corresponding Author: Abdul A.J. Mohamed, Faculty of Science (FOS), Universiti Brunei Darussalam Jalan Tungku, Brunei Darussalam, BE1410, Brunei



drinking, industrial and irrigation purposes (Chen *et al.*, 2002; Mondal *et al.*, 1998). Toxic metals such as arsenic, beryllium, lead and thallium are ubiquitous in nature and could easily find their way into water sources including groundwater (Mendie, 2005). Arsenic and other harmful contaminants in drinking water are a natural calamity and a public health hazard. It exists in the natural systems due to anthropogenic as well as geological sources (Anawar *et al.*, 2003; Ayoob and Gupta, 2006; Rafique *et al.*, 2009; Smedley *et al.*, 2002). The underground water of two sub districts of Tharparkar, Pakistan was found to be severely contaminated with arsenic and other detrimental contaminants (Tasneem *et al.*, 2013).

Metals and other chemicals exist in nature and, even in the absence of anthropogenic activities; their levels might potentially affect the quality of groundwater and pose a risk to human health. A good example is the occurrence of arsenic in Bangladesh groundwater (Anawara et al., 2002). Arsenic as a chemical species disturbs the delicate balance of the ecosystem and affects the living world, which in the long run endangers the health of human beings (Benett et al., 2001). Arsenic in drinking water is carcinogenic if its concentration is above 300 ppb (parts per billion). The latest data indicates that 80% of Bangladesh (40 million people) are at risk of arsenic poisoning-related diseases because the Bangladesh groundwater sources are contaminated with arsenic (Islam et al., 2003).

Acid rain, mine drainage agricultural run-off, industrial and domestic effluents have all contributed to the occurrence of metal loads in the groundwater aquifers. Because metals are non-degradable and thus in the environment they exist as persistent chemical species. In this respect, therefore, metals contaminations in groundwater are of special concern (Bhole and Ramteke, 2011; Chintana, 2002).

Groundwater contamination due to biological, physical and chemical contaminants has become a global problem. Contaminated groundwater is likely a key source of many health problems in Zanzibar Island. This problem has a great secio-economic impact on the domestic water consumers in the Island.

Spring and groundwater are the main sources of domestic water supply in Zanzibar Island, but currently the island is experiencing a severe water shortage. Zanzibar urban-west region is the densiestpopulated region of anzibar Island and there is a continuous increase in water consumption demand. In many parts of the island, large proportion of the population lack access to safe clean water. The water sources are threatened by contamination due to increased urbanization, poor wastewater drainage and other environmental pollution caused by the lack of proper garbage collection and waste disposal.

As there are no drinking water treatment facilities in Zanzibar, it is necessary to monitor frequently the water quality at all sources of domestic water supply to guarantee its safety to the consuming population. Recent study has shown that the levels of hexavalent chromium and copper in some groundwater sources in Zanzibar were beyond WHO recommended limit (Abdul *et al.*, 2013). The monitoring of water quality is thus essential for averting both acute and chronic toxicities. Therefore, the present study is a contribution for the assessment of the occurrence and the levels of As, Be, Pb, Tl and Mg in groundwater from selected domestic water sources in the Zanzibar islands. The alkalinity levels in the studied water sources was also investigated.

2. MATERIALS AND METHODS

2.1. Sampling

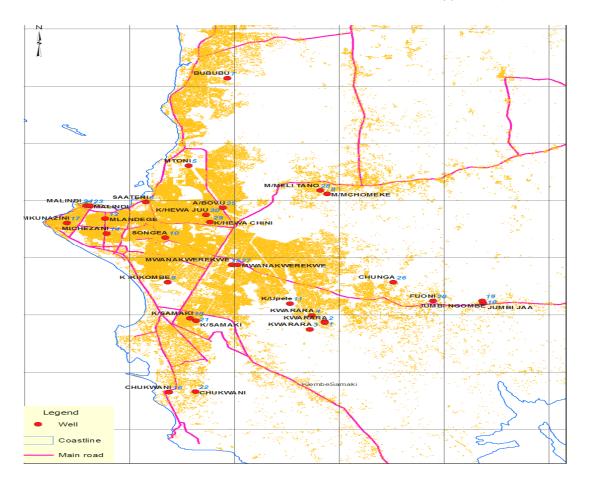
In December 2012, thirty water samples were collected from different areas of the Zanzibar Urban/West region and its territories. Geographic coordinates of all sampling points were recorded using the Global Positioning System (GPS). Sampling positions are shown in **Fig. 1**.

The sources of the collected water samples were; springs, public bore wells, private bore wells, open hand dug wells and closed hand dug wells. Samples were collected in pre-cleaned polyethylene bottles. The handling, storage and preparation of the samples were done in accordance with the standard EPA method 2007.

2.2. Analytical procedure

Palintest photometry procedures were used to determine the levels of the total alkalinity and magnesium, while, the Inductively Coupled Plasma-Optical Emission Spectrometry (ICP-OES) method was used to determine the concentrations of dissolved Arsenic (As), Thallium (Tl), lead (Pb) and Beryllium (Be).





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Fig. 1. Map of sampling sites in Zanzibar Urban-West Region

3. RESULTS

Table 1 shows the levels of alkalinity, magnesium, arsenic, lead, thallium and beryllium in the analyzed water samples. Also presented in **Table 1** are the maximum permissible limits for some of the studied metals as recommended by the US EPA (2009) and WHO (2008). The levels of alkalinity ranged from 38 to 380 mg L⁻¹ as CaCO₃, highest-level was measured at site 23 (Malindi). The alkalinity levels measured at sites 1-4 (Kwarara sampling points) were relatively lower as compared to other areas (**Table 1**). Therefore, water sources in Kwarara areas might have higher degree of corrosivity. Indeed, these sites have been documented for low alkalinity levels (Abdul *et al.*, 2014). Concentrations of Magnesium (Mg) varied from 1.0 to 34 mg L⁻¹. The concentrations of Arsenic

(As) varied from non-detectable (nd) to 6600 ng L^{-1} , Jumbi ngombe had the highest arsenic level. Beryllium (Be) levels ranged from nd to 6100 ng L^{-1} . Indeed, traces of lead (Pb) were present in all water samples with the exception of sampling site 12 (Mlandege). The highest concentration of lead detected was 31400 ng L^{-1} at Thallium (Tl) sampling site 3 (Kwarara). concentrations varied from nd to 11600 ng L^{-1} (Table 1 and 2), site 15 (Mwanakwerekwe) contained the highest Tl level. Anthropogenic environmental pollution due to activities such as improper disposal of wastes, domestic effluents and landfill leachate are the main causes of groundwater contamination in Zanzibar Island. At least one of each of the water sources (except spring water sources) have been contaminated with As, Be, Pb and Tl. Nevertheless, only Beryllium (Be) was not detected in spring water sources (Table 3).

In summary, the proportions of the water samples contaminated with the four hazardous metals were 43.3, 66.7, 70 and 96.7%, for Beryllium (Be), Thallium (Tl), Arsenic (As) and lead (Pb) respectively (**Table 1**). Furthermore, 23.3% of the water samples had Pb

concentrations above the WHO limit. While, 56.67% and 30% are the respective proportions of the water samples with Tl and Be levels beyond the US EPA's maximum limits. All water samples contained arsenic within the WHO concentration limit (**Fig. 2 and 3**).

Table 1. The concentrations of trace metals and levels of alkalinity in the drinking water samples

| | | Alkalinity | Mg | | | | |
|---------------|--------|-----------------------------|---------|----------|----------|----------|----------|
| Sample | Source | (as CaCO ₃ mg/L) | (mg/L) | As(ng/L) | Be(ng/L) | Pb(ng/L) | TI(ng/L) |
| 1 | CHDW | 38 | 7.5 | 0 | 200 | 20800 | 900 |
| 2 | OHDW | 42 | 7.5 | 0 | 500 | 26200 | 0 |
| 3 | OHDW | 65 | 9.0 | 0 | 500 | 31400 | 0 |
| 4 | OHDW | 65 | 4.0 | 0 | 400 | 9600 | 2800 |
| 5 | SW | 228 | 7.0 | 2100 | 0 | 6600 | 6000 |
| 6 | PBW | 240 | 2.5 | 2900 | 0 | 3400 | 4100 |
| 7 | SW | 240 | 8.0 | 470 | 0 | 600 | 430 |
| 8 | PBW | 255 | 6.0 | 480 | 0 | 370 | 5700 |
| 9 | PBW | 183 | 2.5 | 4600 | 0 | 500 | 7100 |
| 10 | CHDW | 130 | 9.0 | 3700 | 0 | 2200 | 8600 |
| 11 | PBW | 205 | 5.0 | 5300 | 0 | 5700 | 9500 |
| 12 | BWP | 330 | 1.0 | 3700 | 0 | 0 | 7800 |
| 13 | CHDW | 200 | 6.0 | 2500 | 0 | 1100 | 7400 |
| 14 | CHDW | 273 | 9.0 | 2500 | 0 | 2500 | 7500 |
| 15 | BWP | 250 | 5.0 | 380 | 0 | 1000 | 11600 |
| 16 | BWP | 260 | 34.0 | 3600 | 0 | 3200 | 9700 |
| 17 | CHDW | 290 | 7.0 | 3900 | 0 | 1200 | 7600 |
| 18 | OHDW | 330 | 7.0 | 4900 | 0 | 3200 | 8400 |
| 19 | OHDW | 228 | 9.5 | 6600 | 0 | 1100 | 9600 |
| 20 | CHDW | 250 | 9.0 | 4900 | 0 | 2900 | 7400 |
| 21 | CHDW | 278 | 8.5 | 4600 | 0 | 900 | 9300 |
| 22 | CHDW | 168 | 9.0 | 0 | 5700 | 4800 | 0 |
| 23 | BWP | 380 | 16.0 | 5100 | 4900 | 12000 | 100 |
| 24 | BWP | 315 | 7.5 | 3000 | 6100 | 15600 | 0 |
| 25 | BWP | 273 | 9.0 | 1000 | 6000 | 15100 | 0 |
| 26 | PBW | 250 | 7.5 | 900 | 5600 | 10800 | 0 |
| 27 | BWP | 188 | 6.0 | 0 | 5500 | 2900 | 0 |
| 28 | OHDW | 135 | 4.0 | 0 | 5500 | 4600 | 0 |
| 29 | OHDW | 142 | 9.5 | 0 | 5600 | 200 | 0 |
| 30 | OHDW | 273 | 7.0 | 0 | 5200 | 4400 | 0 |
| WHO guideline | s* | 50 1 | 10000.0 | | 10000 | | |
| USEPA* | | 1 | 10000.0 | 4000 | 15000 | 2000 | |

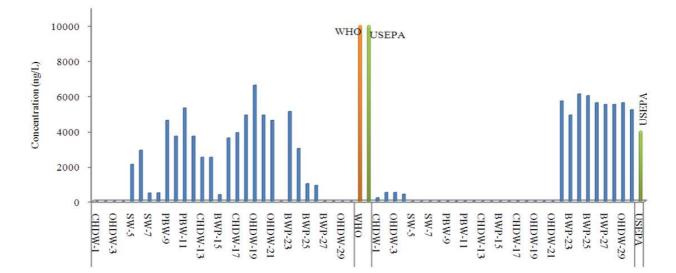
Table 2. Highest concentration/level of the measured water samples' parameters

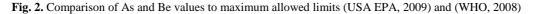
| Parameters | Maximum concentration detected | Water source | Site and name |
|------------------|--------------------------------|--------------|--------------------|
| Magnesium | 34 | BWP | 16(Chukwani) |
| Total alkalinity | 380 | BWP | 23(Malindi) |
| Beryllium | 6100 | BWP | 24(Malindi) |
| Arsenic | 6600 | OHDW | 19 (Jumbi Ng'ombe) |
| Thallium | 11600 | BWP | 15(Mwanakwerekwe) |
| Lead | 31400 | OHDW | 3(Kwarara) |



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| Table 3. Proportions (percentage occurrences) of the measured water samples' parameters | | | | | | | |
|---|------------|------------------|------------|------------|------------|-----------|--|
| Water source | Mg | Total Alkalinity | Be | As | Tl | Pb | |
| | occurrence | occurrence | occurrence | occurrence | occurrence | occurence | |
| | (%) | (%) | (%) | (%) | (%) | (%) | |
| SW | 100 | 100 | NIL | 100.0 | 100.0 | 100.0 | |
| PBW | 100 | 100 | 20.0 | 100.0 | 80.0 | 100.0 | |
| BWP | 100 | 100 | 57.1 | 85.7 | 57.1 | 85.7 | |
| OHDW | 100 | 100 | 66.7 | 33.3 | 44.4 | 100.0 | |
| CHDW | 100 | 100 | 28.6 | 71.4 | 85.7 | 100.0 | |





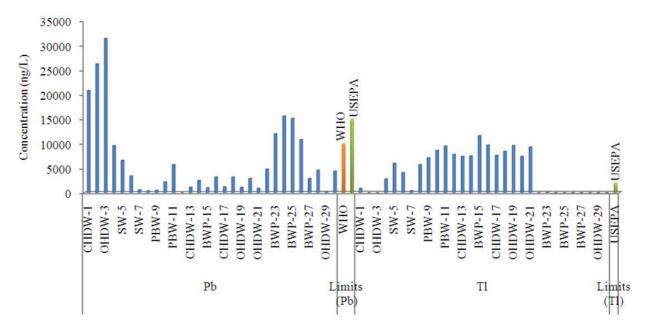


Fig. 3. Comparison of Pb and Tl values to maximum allowed limits (US EPA, 2009) and (WHO, 2008)



4. DISCUSSION

The data obtained in the present study showed that, among the three analyzed carcinogenic elements, arsenic occurred at concentrations below the WHO guidelines for safe drinking water. Although arsenic might occur naturally in the environments, but its occurrence in the studied water sources is most likely of anthropogenic origin. Water site 19 (Jumbi ngombe) had the highest arsenic level, which is possibly a result of arsenic containing fertilizer used at onsite paddy farms, as well as landfill leachate. Some of the water samples had concentrations of lead, beryllium and thallium higher than those set by WHO and the US EPA. Plausible causes for lead occurrence in the majority of water samples could be from natural and anthropogenic sources. Lead is an abundant, naturally occurring metal. Wide application of lead (Pb) in building materials, lead acid batteries, some insecticides and lead-based paints are a possible cause of the broad distribution of lead (Pb) in the water sources and other systems. Moreover, airborne lead entering water sources via precipitation and tetraethyl lead used in gasoline is also among the key contributor of lead contaminants in groundwater aquifers.

The wide distribution of thallium in the Earth's crust is a possible reason for its occurrence in some of the water samples. Nevertheless, the uses of pharmaceutical products, insecticides and rodenticides containing thallium might be a key factor in the significant increase of thallium levels in Zanzibar's groundwater sources. Zanzibar faces growing importation of obsolete electronics, which in turns dumped improperly; therefore, this might be the possible cause for the beryllium contamination in the groundwater bodies. In addition, the poor drainage systems in Zanzibar cause contaminants to be carried by runoff from one point to another.

5. CONCLUSION

Some of the groundwater sources in Zanzibar Island are contaminated with trace amounts of As, Be, Pb and Tl. The levels of Be Pb and Tl exceeded the accepted international limits for safe drinking water. In this regard, prompt action is required to mitigate this apparent health risk. The present study provides a baseline data and information for current and future domestic water quality monitoring programs in Zanzibar.

Because of the time constraint, this study focused only on the selected metals at selected water-sampling points. Therefore, further studies on assessing water quality in other regions of the Island are needed because the level of contamination may vary from one area to another. The present study could not look at the epidemiological association of these metals and Zanzibar cancer patients. There is also a need to undertake epidemiological studies about the presence and the levels of carcinogenic metals from patients with cancer cases. It is also important to establish drinking water treatment plants at different locations of the Island; this can help to monitor different water parameters on daily basis.

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