

Development of Nanostructured Stress Free Pt-Rich FePt Films for Micro Electro Mechanical System Applications

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Abstract: Problem statement: The electroplating technique is especially interesting due to its low cost, high throughput and high quality of deposit. Magnetic thin films are extensively used in various electronic devices including high density recording media and Micro Electro Mechanical (MEMS) devices. So we examined FePt films and give good magnetic properties. **Approach:** The Pt-rich FePt alloys were electrodeposited galvanostatically by various temperature baths. **Results:** The elemental percentage of Pt in the film was very high determined by X-ray spectroscopy. Surface characterization was carried out by using X-Ray Diffractometer (XRD) and Scanning Electron Microscope (SEM). The magnetic properties determined by Vibrating sample magnetometer. Hardness and adhesion of the films were also discussed. **Conclusion/Recommendation:** The films were soft magnetic character of lower temperature bath and become a hard magnetic character of higher temperature bath. Therefore this magnetic films could be an important material for incorporate in Micro electro mechanical devices.

Key words: FePt, electrodeposition, magnetic film, magnetic properties

INTRODUCTION

With the progress in the field of Micro Electro Mechanical System (MEMS) technologies^[1-5] there has been growing interest in developing electroplated, nanostructured soft and hard magnetic materials^[6,7] for microactuators, micromotors and microswitches. The possibilities of these electroplated materials, retaining hard and soft magnetic properties up to several microns thickness, gives researches opportunity to explore them for micro fabrication of MEMS devices. Recently, much effort is being made to electrodeposits also materials of the group of L10 ordered alloys, like FePt^[8,9] and CoPt^[10,11], because they exhibit a significantly higher uniaxial magneto crystalline anisotropy. As the formation of the L10 phase is kinetically hindered at room temperature, post annealing of the films is necessary. Electrodeposited and post annealed FePt and CoPt films can reach coercivities exceeding 1T^[11].

In the present study we investigated in detailed the effects of bath temperature on electrodeposited Pt rich magnetic FePt films of various deposited time and current density. Also we discussed their structural and magnetic characterization.

MATERIALS AND METHODS

A copper substrate of size 1.5×5.0 cm as cathode and pure steel of same size as anode were used for galvanostatic electrodeposition experiments. Current for electrodeposition was passed from a regulated directed current unit. Analytical reagent grade chemicals were used to prepare baths. An adhesive tape was used to mask off all the substrate except the area on which deposition of film was desired. Each substrate was buffed for removing scratches in a mechanical polishing wheel using a buffing cloth coated with aluminum oxide abrasive. Buffed substrates were decreased using acetone. Before electrodeposition these substrates were electrocleaned in an alkaline electrocleaning bath. The bath contained sodium hydroxide: 7.0 g L⁻¹; sodium carbonate: 20.0 g L⁻¹; trisodium phosphate: 9.0 g L⁻¹ and sodium metasilicate: 24.0 g L⁻¹. The bath was operated at 3.0A dm⁻². After electrocleaning the substrates were rinsed in distilled water. FePt films were electrodeposited on polycrystalline Cu substrate from a single bath containing 1m H₂PtCl₆, 0.1 (NH₄)₂SO₄ and 0.1 mol L⁻¹ FeSO₄. The solution pH was adjusted to 3 by adding a small amount of either sulfuric acid or hydrochloric acid. For depositing films using dc plating at current

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densities varying from 1-3 mA cm⁻² at various temperature 30, 50, 70°C with various time of deposition.

The thickness of the deposits was tested using digital micrometer (Mitutoyo, Japan). Magnetic properties of deposited films were studied using vibrating sample magnetometry. In this technique the material under study was contained in a sample holder, which was centered in the region between the pole pieces of a laboratory magnet. A slender vertical sample rod connects the sample holder with a transducer assembly. The transducer converts a sinusoidal alternating current drive signal into a sinusoidal vertical vibration of the sample rod. Coils mounted on the pole pieces of the magnet pick up the signal resulting from the sample motion. X-Ray Diffractometry (XRD) and Scanning Electron Microscopy (SEM) were used to study the structure and morphology of these magnetic films respectively. From XRD data crystallite size of the deposited CoNiP and film stress were calculated. Percentage of elements such as cobalt, nickel and phosphorous present in the deposits were obtained as follows. For elemental analysis CoNiP film was electrodeposited on stainless steel substrate to ensure easier peeling-off of the film. After deposition the film was peeled off from the substrate. It was dissolved in 3:1v/v of H₂SO₄ and HNO₃ and the percentage composition was obtained using energy dispersive X-Ray Spectroscopy (EDS). Hardness of the deposit was obtained using Vicker's hardness tester using diamond indenter method. Adhesion of the film was tested by bend and by scratch or chisel test. These tests are widely used in the field of electroplating^[12].

RESULTS

Surface characterization: X-ray diffraction patterns of various FePt electrodeposits produced from various temperature bath like 30, 50 and 70°C for 1.0 mA cm⁻² and 30 min were fixed current density and time of deposition respectively for electrodeposition (Fig. 1). The data obtained from the XRD pattern compared with standard data and were found to have face centered tetragonal structure and exhibited (221) plane predominantly. From the XRD pattern peak, stressed in the film was calculated using the formula: Young's modulus = stress/strain. The results shown in Table 1.

Crystalline size of the deposits were calculated from the XRD pattern using the formula:

Crystalline size = $0.9\lambda/B\cos\theta$. These values clearly show that the crystallite size of the FePt deposit obtained by electrodeposition process are in the nano

scale. The crystallite size of deposits are shown in Table 1.

Table 1: Effect of bath temperature on the structural and mechanical properties of FePt film electrodeposited at 1 mA cm⁻² for 30 min

Bath temperature (°C)	Crystalline size (nm)	Vickers hardness (VHN)	Internal stress (MPa)	Film composition (at%)
30	76	268	163	95.0
50	22	275	152	96.4
70	18	317	140	97.9

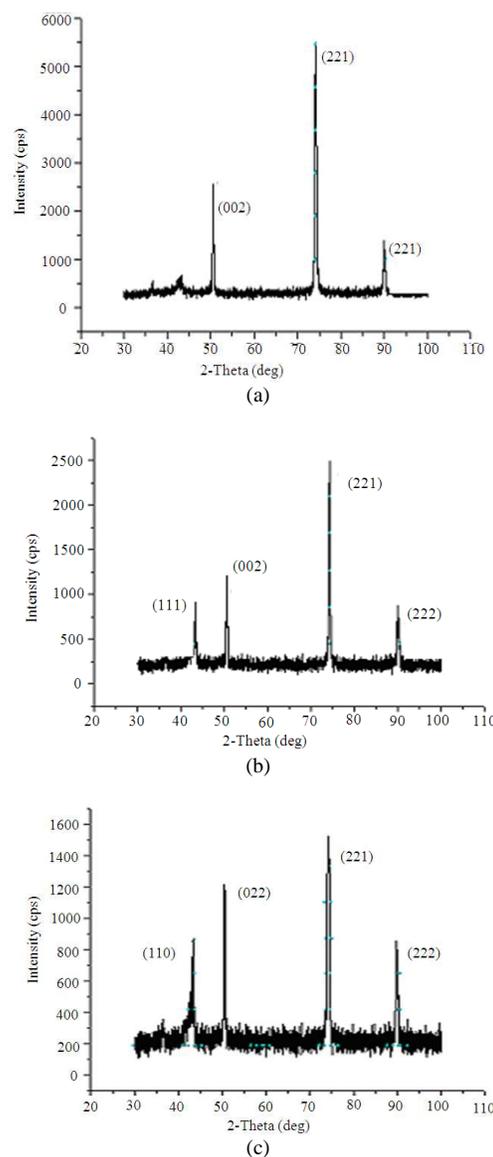


Fig.1: XRD patterns for electrodeposited FePt film at current density: 1mA cm⁻², deposition time 30min and various temperatures bath (a): 30°C, (b): 50°C, (c): 70°C

Table 2: Effect of bath temperature at 30°C, time of deposition and Current density on the thickness and magnetic properties of electrodeposited FePt film

Current density (mA cm ⁻²)	Time of deposition (min)	Thickness of deposit (um)	Magnetic saturation (emu)	Remanent (emu)	Coercivity (Oe)	Squareness
1	15	3.0	0.733	0.013	500	0.017
	30	3.2	0.718	0.019	600	0.026
	60	3.5	0.715	0.024	700	0.033
2	15	3.5	0.720	0.034	600	0.047
	30	4.0	0.700	0.037	700	0.052
	60	4.2	0.680	0.039	800	0.057
3	15	5.0	0.650	0.045	700	0.069
	30	5.3	0.620	0.050	800	0.080
	60	5.6	0.600	0.052	900	0.086

Table 3: Effect of bath temperature at 50°C, time of deposition and current density on the thickness and magnetic properties of electrodeposited FePt film

Current density (mA cm ⁻²)	Time of deposition (min)	Thickness of deposit (um)	Magnetic saturation (emu)	Remanent (emu)	Coercivity (Oe)	Squareness
1	15	3.0	0.427	0.024	700	0.056
	30	3.2	0.418	0.025	800	0.059
	60	3.5	0.400	0.028	900	0.070
2	15	3.5	0.420	0.030	700	0.070
	30	4.0	0.400	0.033	800	0.080
	60	4.2	0.380	0.037	900	0.097
3	15	5.0	0.350	0.040	900	0.114
	30	5.3	0.320	0.042	1000	0.134
	60	5.6	0.300	0.043	1100	0.140

Table 4: Effect of bath temperature at 70°C, time of deposition and Current density on the thickness and magnetic properties of electrodeposited FePt film

Current density (mA cm ⁻²)	Time of deposition (min)	Thickness of deposit (um)	Magnetic saturation (emu)	Remanent (emu)	Coercivity (Oe)	Squareness
1	15	3.0	0.233	0.045	1000	0.193
	30	3.2	0.218	0.048	1100	0.220
	60	3.5	0.215	0.050	1200	0.232
2	15	3.5	0.214	0.054	1200	0.252
	30	4.0	0.212	0.057	1300	0.268
	60	4.2	0.200	0.059	1400	0.295
3	15	5.0	0.150	0.065	1500	0.433
	30	5.3	0.120	0.067	1600	0.558
	60	5.6	0.100	0.069	1700	0.690

Electrodeposited FePt films from all three temperature bath as mentioned in XRD studies were subjected to SEM studies. The micrographs are shown in Fig. 2. In general, microstructure of the FePt film is greatly influenced by the temperature of bath. Elements present in the film were analyzed by Energy dispersive X-ray spectroscopy and the results are shown in Table 1.

Mechanical properties: Adhesion of the film with the substrate is tested by bend test and scratch test. It showed that the film is having good adhesion with the substrate. Hardness of the film increases when the film deposited from bath temperature increases. This may also be due to the lower stress associated with FePt film. The results are reported in Table 1.

Magnetic properties: Electrodeposition studies were carried out using different bath temperature. Table 2

shows the results of electrodeposition of FePt and their magnetic properties. The temperature of the bath varies at 50°C and electrodeposition studies were carried out. The thickness of the film for various current densities and time of deposition as shown in Table 3.

Table 4 shows the effect of deposited film by higher bath temperature 70°C. The thickness of deposit increases with increase in current density. The change in magnetic properties was because of the higher percentage of Pt in the films make low stress present in the film.

DISCUSSION

From XRD data the crystalline structure found in FePt film is fct produced from a bath at low temperature has high stress and this is due to random crystal orientation during electrodeposition. But on the

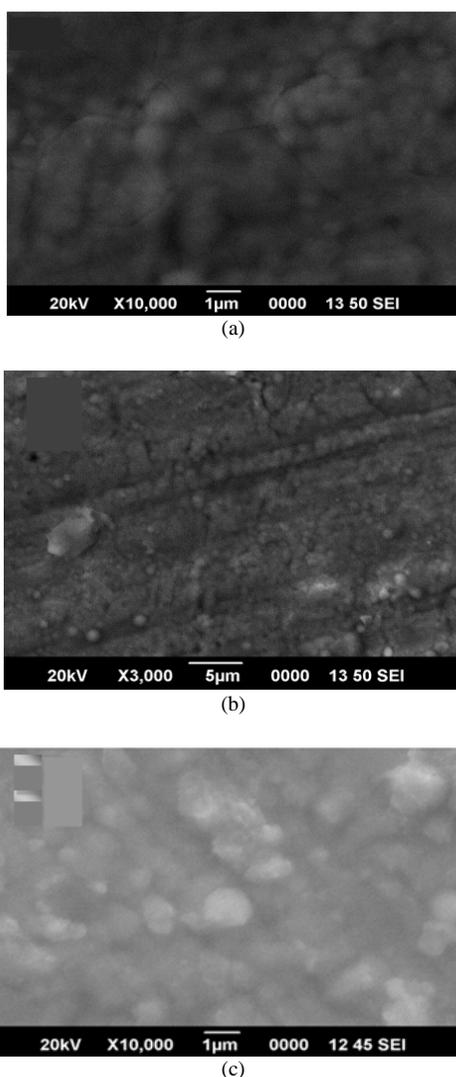


Fig. 2: SEM images of FePt films electrodeposited for 30min, Current density: 1 mA cm^{-2} at temperature (a): 30°C , (b): 50°C , (c): 70°C

film from higher temperature bath has low stress. This is due to uniform crystal orientation during electrodeposition. This analysis reveals the effect of film from higher temperature bath on the crystallite size of the deposit. The morphology of the film deposited at low temperature bath appeared crevice pattern with crack, grains are formed in random order and it also appears less bright. In the film deposited at high temperature, the grain sizes are visible and granular form. They are in more order with fct structure.

All deposits, which were subjected to analysis, EDS have greater than 95% Pt content. The films obtained from various temperatures and current

densities are Pt rich films having high magnetic properties. The films are high hardness and good adhesion.

When the low temperature bath (30°C) used for deposition, the thickness of the deposit increases with increase in current density and time of deposition. Films are dull in appearance and having pits. Films are having very low coercive and remanent values. In the electrodeposition studies films produced (50°C) were uniform and bright. This because the orientation of crystallization uniformly during electrodeposition by adsorbing itself on the initially deposited crystals. The magnetic properties of the high temperature bath (70°C) films revealed that these films are having a high coercive and low remanent value when compared to the deposit obtained from films deposited from low temperature bath.

As the average crystallite size of the films are in nanoscale, considerable change in magnetic behavior can occur. When the crystalline size approaches high nanolevel, the domain wall thickness is comparable to the crystalline size the coercivity found to increase. Analysis of crystallite size, microstructure and magnetic properties confirm that the origin of magnetic properties is because of the strongly interacting array of single domain crystals. This is mainly due to the films from high temperature bath.

CONCLUSION

A FePt film having good hard magnetic properties can be electrodeposited from the high temperature bath containing percentage of Pt is rich. In the low temperature bath the film character is change it showed soft magnetic character. It also increases the film stress, which is a cause for cracked film. The high temperature bath films are lower stress which are used in MEMS devices. Hardness of the film also increase in higher temperature bath films which are having high coercivity value and low remanent values. Also these films have good adhesion with the substrate and their crystalline sizes are nano scale.

REFERENCES

1. Cugat, O. *et al.*, 2003. Magnetic micro-actuators and systems. IEEE. Trans. Magnet., 39: 3607. DOI: 10.1109/TMAG.2003.816763
2. Roy, S. *et al.*, 2005. Pulse reverse plating for integrated magnetic on Si. J. Magnetism Magnet. Mater., 1524: 290-29. DOI: 10.1016/j.jmmm.2004.11.566
3. Rhen, F.M. *et al.*, 2005. Thick film permanent magnets by membrane electrodeposition. J. Applied Phys., 97: 113908-113908-4. DOI: 10.1063/1.1923587

4. Lee, K.H. *et al.*, 2002. Magnetic properties of self-ordered ferromagnetic nano wires by ac electroforming. *J. Applied Phys.*, 91: 8513. <http://link.aip.org/link/JAPIAU/91/8513/1>
5. Emerson, R.N. *et al.*, 2007. Effect of organic additives on the magnetic properties of electrodeposited Conip hard magnetic films. *Thin Solid Films*, 515: 3391-3396. DOI: 10.1016/j.tsf.2006.09.034.
6. Pattnaik, G. *et al.*, 2006. Electrodeposition of hard magnetic films and microstructures. *J. Elect. Acta*, 852: 2755-2764. DOI: 10.1016/j.electacta.2006.07.062
7. Park, H.D. *et al.*, 2006. Microstructure and magnetic properties of electrodeposited CoPtP alloys. *J. Applied Phys.*, 99: 08N305-08N305-3. DOI: 10.1063/1.2173229
8. Leistener, K. *et al.*, 2004. Phase formation, microstructure and hard magnetic properties of electrodeposited FePt films. *J. Applied Phys.*, 95: 7267-7269. DOI: 10.1063/1.1667438
9. Moffat, T.P. *et al.*, 2008. Oxygen Reduction Kinetics on Electrodeposited Pt, Pt_{100-x}Ni_x and Pt_{100-x}Co_x. *Electrochem. Solid State Lett.*, 156: B238-B251. <http://dx.doi.org/10.1149/1.3033515>
10. Eagleton, T.S. *et al.*, 2005. Electrodeposition of CoxPt1-x thin films. *J. Electrochem. Soc.*, 152: C27-C31. DOI: 10.1088/0022-3727/38/6/020
11. Leistener, K. *et al.*, 2003. Highly coercive electrodeposited FePt films by postannealing in hydrogen. *Appl. phys. Lett.*, 85: 3498-3500. <http://link.aip.org/link/?APPLAB/85/3498/1>