Antimicrobial Effects of 1-Monocaprylin and 1-Monocaproin Through *in vitro* Growth Inhibition and Molecular Docking Studies

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Abstract: 1-monocaproin and 1-monocaprylin synthesized through a novel process by the chemical reaction of glycidol and their respective fatty acids with copper acetate as the catalyst possessed the amphiphilic property. 1monoacyl glycerols synthesized were found to exhibit antibacterial, antifungal, anti parasitic and antiviral properties. 1-monocaproin and 1monocaprylin were able to show their antibacterial effect by inhibiting the growth of the Gram negative bacteria Escherichia coli, Pseudomonas aeruginosa and Gram positive bacteria Staphylococcus aureus, Bacillus subtilis at concentrations of 100-500 ppm. The Minimal Inhibitory Concentration (MIC) of both 1-monoacyl glycerols were found to be 0.5 ppm. 1-monocaproin and 1-monocaprylin were able to show their antifungal effect by inhibiting the growth of the filamentous fungi Mucor racemosus and Rhizopus stolonifer at the concentration of 1000 ppm. Based on the molecular interaction and common binding interaction study, 1monocaproin is expected to exhibit a similar antiviral activity as that of Oseltamivir to H₅N₁ influenza virus hemagglutinin. 1-monocaprylin and 1monocaproin synthesized using copper acetate could exhibit a broad spectrum antimicrobial effect in combination with other monoacyl glycerols or with other antimicrobial agents.

Keywords: Monoacyl Glycerols, 1-Monocaprylin, 1-Monocaproin, Antimicrobial, Anti Bacterial, Antifungal, Antiviral

Introduction

Food-borne diseases caused by the microbial contamination from bacteria, fungi, parasite and virus pose major public health problems in developed and developing countries (Altekruse et al., 1999). Food-borne infectious diseases cause serious day to day problem for the health care system and led to tremendous economic loss and hence it is important to develop the means to control the transmission from food to humans (Newell et al., 2010). This is achieved by either limiting or preventing the growth of undesirable microbial flora in food products (Dolezalkova et al., 2012). More new efficient methods for the decontamination or prevention of food contamination by these pathogens would therefore be desirable. Among the viral borne illness, avian influenza such as A (H₅N₁) and A (H_7N_9) are the ones which caused serious infection in human beings. two surface The glycoprotein;

Hemagglutinin (HA) and Neuraminidase (NA) play a vital role in the attachment and release of the Influenza A virus, respectively (McNicholl and McNicholl, 2001; CDC, 2006; Lupiani and Reddy, 2009). Hemagglutinin (HA), the glycoprotein in influenza virus envelope, plays a critical role in viral binding, fusion and entry processes. Therefore, HA is a promising target for developing anti-influenza drugs, which block the initial entry step of viral life cycle. Recently, molecular modeling and computational chemistry based computer-aided drug design provided great help for modern drug development (Marshall, 1987). Software programs such as Auto Dock were widely used to search potential inhibitor for protein targets (Morris, et al., 2009). In this study, we employed the strategy of molecule docking to explore 1-monocaproin comparing with Oseltamivir as potential H₅N₁ hemagglutinin inhibitors.

Monoacyl glycerols (MAGs) are gaining more attention as antimicrobial agents nowadays as they are generally



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considered as non-harmful safe agents with no adverse reactions and non- toxic to mucosa which makes them suitable for wide industrial applications. MAGs exhibit inhibitory effects against major food-borne microorganisms due to their amphiphilic and emulsifying properties (Jan et al., 2003). The precise mode of the action against microorganisms is not yet clear and postulated several hypotheses. One hypothesis is based on the amphiphilic character of MAG molecule which aids in their penetration and incorporation into cytoplasmic membrane and subsequently disrupts the cell permeability and prevents the transportation of the nutrients (Nair et al., 2004; Dufour et al., 2007). An alternative hypothesis postulates the penetration of fatty acids and dissociated in the inner cell environment, thereby increasing the intracellular acidity (Sun et al., 1998). The physicochemical and functional properties as well as antimicrobial activity of MAGs depend on the number of carbon atoms and double bonds present in the fatty acid chain (Wang and Johnson, 1997). There are many studies published recently for the monoacylglycerol with C-10:0 and above showing the inhibitory effects on the growth of microorganisms (Kabara et al., 1977; Razavi-Rohani and Griffiths, 1994). Hence, MAGs with C-6:0 and C-8:0 were chosen for our current study with the objectives; (1) synthesis of 1-mono caproin and 1-monocaprylin by a novel process through the catalytic addition of respective fatty acids to glycidol with copper acetate as the catalyst; (2) evaluation of the inhibition property against selected food borne bacteria and fungi through in vitro and MIC studies and (3) evaluation of the antiviral property of 1-monocaprion through the molecular interaction study and common binding interaction with crystal structure of H₅N₁ influenza virus hemagglutinin as drug target protein along with a standard drug Oseltamivir.

Materials and Methods

Details of Microorganisms

Bacterial Strains

Salmonella typhimurium (ATCC 23765), Bacillus subtilis (ATCC 25912), E. coli (ATCC 25922), Staphylococcus aureus (ATCC 25923), Pseudomonas aeruginosa (ATCC 27853), Sterptococcus pneumonia (ATCC 34501).

Fungal Strains

Aspergillus niger (ATCC 22342), *Rhizopus oryzae* (ATCC 9363), *Aspergillus flavus* (ATCC 9643) and *Leucothrix mucor* (ATCC 27891).

Monoacylglycerols (MAGs)

MAGs with C-06:0 and C-08:0 were prepared as per Janis et al. (2000) published earlier but by a modified procedure using copper acetate as the catalyst. Initially, the catalysts were prepared by the reaction of Cu(AcO)₂.H₂O with hexanoic acid and octanoic acid for the synthesis of 1- mono caproin and 1- mono caprylin respectively with toluene as the reaction medium. 1mono caproin was prepared with 0.02 mol hexanoic acid with 0.25g of catalyst prepared with hexanoic acid with 0.03 mol of glycidol and the yield was 65%. 1- mono caprylin was prepared with 0.01 mol octanoic acid with 0.1g of catalyst prepared with octanoic acid with 0.015 mol of glycidol and the yield was 60%. Products were purified for the removal of residual glycidol and to reduce the quantity of copper ions from the catalyst through column chromatography. MAGs were run through Thin Layer Chromatography (TLC) to determine their purity and confirmed by Nuclear Magnetic Resonance spectroscopy (NMR) and ensured the residual copper ions to be $<50 \text{ mg kg}^{-1}$.

Assay of Antibacterial Activities

Bacterial strains were maintained on nutrient agar until use. The antibacterial activity of 1-monocaprion (HAP) and 1-monocaprylin (OAP) were screened using agar well diffusion method (Jahangirian *et al.*, 2013) against the selected organism mentioned in section 2.1.1 along with control at concentrations of 100, 250 and 500ppm and the zone of inhibition was measured with antibiotic zone scale in mm.

Minimum Inhibitory Concentration (MIC)

MIC of the compounds OAP and HAP were determined by agar diffusion assay method (Jahangirian *et al.*, 2013) at four different concentrations (0.5, 1, 10 and 100ppm each) against *E. coli* and *S. aureus* along with control in duplicate samples.

Assay of Antifungal Activities

1-monocaprion (HAP) and 1-monocaprylin (OAP) were screened for antifungal activity (Serrano *et al.*, 2004) against the selected organism mentioned in section 2.1.2 at the concentration of 1000ppm along with control with potato dextrose agar and the inhibition was represented as percentage

Preparation of Molecular Structure

To perform the molecular docking between hemagglutinin and potential inhibitors, we employed the Crystal Structure of a H_5N_1 avian influenza virus hemagglutinin of PDB: 2KF0. The chemsketch tools were used to draw the ligands and converted in to protein data bank format for the docking using the chemical language format.

Molecular Docking

In this study, we employed Auto Dock 4.02 v software for the preparation of drug target protein, The polar hydrogen's were added along with kollmann charges and the respective drug target protein was saved in current mode of protein data bank. Docking was carried out with genetic algorithm mode with population size of 150, with the maximum number of energy evaluations of 2500000, maximum numbers of generations being 27000 with 20 runs. The best run with negative binding energy was considered as best results. The interaction results were visualized using acclerys discovery studio 4.5 visualizer. Using Auto Grid program the interaction energies for the ligand, standard drug and the drug target protein were calculated and based on the above, the grid was placed in equal dimensions $50 \times 50 \times 50$ in XYZ dimensions with 0.4 spacing in angstroms along box is placed in -74.536 X, 12.015 Y, 33.301 Z directions respectively. Since the active sites of hemagglutinin in H_5N_1 were highly conserved, the docking pocket were compared with ten key amino acid residues including PHE B:63, ALA B:65, PHE B:88, LEU B:89, TRP B:92, GLY A:304, GLY A:303, CYS A:305, PRO A:306 and TYR A:308 and compared with Oseltamivir and 1-monocaproin.

Results and Discussion

Anti-Bacterial Effect

Antibacterial effects of both tested MAGs as Zone of inhibition are arrayed in Table 2 and the Zone of inhibition represented in Fig. 2. Out of the three Gram negative bacteria tested, both 1-monocaproin (HAP) and 1-monocaprylin (OAP) were effective and shown inhibition at a concentration of 100ppm itself for *Escherichia coli* and *Pseudomonas aeruginosa*. However, HAP and OAP were not inhibitive to *Salmonella typhimurium*. Out of the three Gram positive bacteria tested, both 1-monocaproin and 1monocaprylin were effective and shown inhibition at 100ppm itself for *Staphylococcus aureus* and *Bacillus subtilis*. However, both the monoglycerides were not inhibitive to *Sterptococcus pneumonia*.

There were lot published literatures to prove the inhibitory property of various MAGs against Grampositive bacteria (Kabara *et al.*, 1972; Schlievert *et al.*, 1992; Oh and Marshall, 1993; Branen and Davidson, 2004; Bunkova *et al.*, 2011) and the resistant property to Gramnegative bacteria (Skrivanova *et al.*, 2006; Kabara, 1978; Kabara, 1984). Based on detailed published studies of saturated fatty acids with chain lengths between 6 and 18

carbons, the lauric acid was found to possess more inhibitory activity against Gram-positive bacteria (Kabara *et al.*, 1972) and MAG synthesized with caproic acid and caprylic acid in our study were found to show similar inhibitory effect.

In the earlier studies, *E. coli* and *P. aeruginosa* were proved to be show resistance against monocaprin even at the concentration of 1,000 mg L⁻¹. However, in our studies, both 1-monocaproin and 1-monocaprylin were found to be inhibitive at 0.5 ppm itself for these two species and to the Gram negative bacteria *Staphylococcus aureus* and *Bacillus subtilis* as shown in Fig. 3. Similar inhibitory effects were observed in monolaurin along with lactic acid on *Staphylococcus aureus* on meat products (Kabara, 1985). Monocaprin was found to increase the sensitivity of *Bacillus cereus* spores in thermal inactivation (Chaibi *et al.*, 1998) and combined with monolaurin inhibited the growth of *Listeria monocytogenes* (Nair *et al.*, 2004).

Antifungal Effect

Similarly, the MAGs were tested against fungi; Aspergillus niger, Rhizopus oryzae, Aspergillus flavus and Leucothrix mucor. Antifungal effects of both tested MAGs as % of inhibition are expressed in Table 3. Monocaproin had shown 60% inhibition against filamentous fungi, Mucor racemosus and Rhizopus stolonifer whereas Monocaprylin had shown 80 and 65% inhibition respectively. Both the MAGs were non inhibitory to A. niger and A flavus.

Mucormycosis, caused by Mucor species are the second prevalent mucoralean fungus surpassing the Rhizopus (Petrikkos al., species et 2012). Monoglycerides had also found to prevent or inhibit the growth of yeasts and filamentous fungi (Bergsson et al., 1998; Bunkova et al., 2010; Ruzicka et al., 2003). Monocaprin was found to be most efficient against all tested Gram-positive bacteria, yeasts and filamentous fungi except Mucor racemosus (Dolezalkova et al., 2012; Bergsson et al., 2001) whereas our compound monocaprylin showed good inhibition against Mucor racemosus and Rhizopus stolonifer. Monolaurin had shown the inhibition of spore at concentration of 0.5 mg per ml for A niger and Penicillium (Rihakova et al., 2001; Mansour et al., 1996) however, the compounds tested in this study were non inhibitory.

Antiviral Property

In this study, 1-monocaproin along with Oseltamivir was employed for molecular docking with H_5N_1 hemagglutinin as shown in Fig. 1 in which the HA_0 glycoprotein of influenza virus is an integral

membrane protein (type Ι transmembrane glycoprotein) with approximate dimensions of 135 Å (length) \times 35-70 Å (radius). Each monomer of the HA molecule consists of a globular head domain and a stem domain. The globular domain consists of a part of HA1 only (including the receptor-binding domain and vestigial esterase domain), whereas the stem domain contains parts of both HA₁ and HA₂ which mediates the viral adsorption, membrane fusion, thus realizing influenza virus entry. The potential inhibitor property of 1-monocaproin was compared with Oseltamivir. The general molecular properties of the target test compounds Oseltamivir and 1-monocaproin are detailed in Table 1. The conformation variations, binding energies and inhibition constants for the molecular docking of 1-monocaproin and Oseltamivir

were presented Table 4. The minimum binding energy and other properties revealed that the 1-monocaproin and Oseltamivir could be successfully docked. The estimated free energy of 1-monocaproin was - 4.39 kca/mol and Oseltamivir was -7.44 kca/mol. 1monocaproin showed almost 60% equivalent free energy binding ability to HA when compared to standard drug Oseltamivir. Table 5 shows the interaction of the amino acids in the H₅N₁ HA pocket and compared with Oseltamivir and 1-monocaproin for its common interaction. Oseltamivir a standard drug showed interaction with ten amino acids with the drug target protein and 1-monocaproin showed interaction with five amino acids. The amino acid residue; PHE B: 63, TYR A: 308, CYS A: 305 showed common binding interaction in both compounds as represented in Fig. 4 and Table 5.



Table 2. Thillbacterial effects of monoacyigiyeerois with 6 C-00.0 and with 6 C 00.0
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	Bacteria character	Zone of Inhibition (mm)						
		MAG C06:0 - HAP			MAG C-08:0 – OAP			
Bacterial strain		500ppm	250ppm	100ppm	500ppm	250ppm	100ppm	
Escherichia coli	Gram-negative	10	11	12	8	10	11	
Salmonella typhimurium	Gram-negative	NI	NI	NI	NI	NI	NI	
Pseudomonas aeruginosa	Gram-negative	10	10.75	11	10	11.75	12.45	
Staphylococcus aureus	Gram-positive	8	9	9	10	10.25	11	
Bacillus subtilis	Gram-positive	8	8.5	9	10	10.5	10.5	
Sterptococcus pneumonia	Gram-positive	NI	NI	NI	NI	NI	NI	

Values represent the average of duplicate samples; NJ -Not Inhibiting; HAP- Monocaproin; OAP - Monocaprylin

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Table 3. Zone of Inhibition of monoac	ylglycerols MAG C-06:0 and	MAG C-08:0 against filamentous fur	ngi	
	% of inhibition			
Microbial strain	Fungal character	MAG C-06:0 - HAP	MAG C-08:0 - OAP	
Aspergillus flavus	Filamentous fungi	NI	NI	
Aspergillus niger	Filamentous fungi	NI	NI	
Mucor racemosus	Filamentous fungi	60	80	
Rhizopus stolonifer	Filamentous fungi	60	65	
Values represent the average of duplic	ate samples; HAP- Monocapro	oin; OAP – Monocaprylin; NI - Not I	nhibiting	
Table 4. The Energies of binding betw	een H ₅ N ₁ HA and the targeted	compounds 1-monocaproin and Ose	ltamivir	
Parameters	1-monoca	proin (Drug candidate)	Standard oseltamivir	
Estimated inhibition constant, Ki	98.61µm		3.52 µm	
Estimated free energy of binding	- 4.39 kca/mol		-7.44 kca/mol	
Final intermolecular energy	-6.77 kca/mol		-9.83 kca/mol	
vdW + Hbond + desolv energy	-6.61 kca/mol		-9.78 kca/mol	
Electrostatic energy	-0.17 kca/mol		-0.04 kca/mol	
Final total internal energy	-0.57 kca/mol		-1.11 kca/mol	
Torsional free energy	+2.39 kca	/mol	+2.39 kca/mol	
Unbound system's energy	-0.57 kca	mol	-1.11 kca/mol	
Table 5 Dir ding an anal of the lines d		hamaaalutinin (IIN) dmaataarat af	4	
Table 5. Binding energy of the figands	s to the active site/binding site	$\frac{1}{1} + \frac{1}{1} + \frac{1}$		
Hemagglutinin (H_5N_1) drug target	Amino	acid binding	distance in A	
1-Monocaproin	PHE B	:63	1.97	
	HIS A:	295	4.41	
	TYR A	.:308	5.26	
	THR A	.:301	2.48	
	CYS A	.:305	3.51,2.18	
Oseltamıvır	PHE B	:63	4.21,2.16,2.18	
	ALAE	8:65	3.55	
	PHE B	:88	4.79	
	LEU B	:89	4.85	
	IRP B	:92 -204	2.76	
	GLY A		2.84	
	GLY A		2.88	
		.303	2.03	
	PRO A		2.75	

Note: A and B are chain in proteins



CYS A:305 PRO A:306 TYR A:308

1.96

Fig. 1. Structural features of the influenza virus hemagglutinin. The picture shows α-helices and β-sheets of HA₀ folded into a compact monomer; an unfolded polypeptide chain of H_5N_1 virus, which is composed of Heamagglutinin (HA) containing 3 different regions: (i) fusion (comprised of the N- and C-terminal regions of HA1 (F' sub-domain) and HA2 (F sub-domain), which are located in the stem domain, a main structure responsible for membrane fusion machinery), (ii) esterase (comprised of two sub-domains between Fusion and receptor binding regions, which are located in the middle of folded HA and (iii) receptor binding (almost in the middle of HA₁ of unfolded HA and at the top of folded HA regions)

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Fig. 2. Zone of inhibition. 1-monocaproin (HAP) and 1-monocaprylin (OAP) showing anti bacterial effect at a minimal dosage of 100 ppm against Gram negative bacteria *Escherichia coli* and *Pseudomonas aeruginosa* and gram positive bacteria *Staphylococcus aureus* and *Bacillus subtilis*



Fig. 3. Minimum Inhibitory effect 1-monocaproin (HAP) and 1-monocaprylin (OAP) showing MIC at the concentration of 0.5 ppm for *Staphylococcus aureus* and *Bacillus subtilis*

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Fig. 4. Active site amino acid with 1-monocaproin and Oseltamivir; Binding interaction

Nowadays, two classes of anti-influenza drugs, M₂ ion channel inhibitors and neuraminidase inhibitors respectively, are used for prophylaxis and treatment of influenza virus infection. Hemagglutinin (HA), the glycoprotein in influenza virus envelope, plays a critical role in viral binding, fusion and entry processes. Therefore, HA is a promising target for developing antiinfluenza drugs, which block the initial entry step of viral life cycle (Shen et al., 2013). Generally, AIV including H₅N₁ are sensitive to Oseltamivir and a small number of H₅N₁ strains isolated from avian and human origin have been reported to exhibit resistance to Oseltamivir. Oral application of Oseltamivir via drinking water reduced the morbidity, mortality, virus excretion and chicken-to-chicken transmission in HPAIV H₅N₂ experimentally infected chickens (WHO, 2012). Oseltamivir was non-toxic for chicken embryos and prevented the replication of an HPAIV H7N1 in inoculated eggs (Allen et al., 2006). Since Oseltamivir had shown strong interaction with the amino acids in the active site of Hemagglutinin of H₅N₁ virus, it is considered to block the HA-mediated viral entry by interfering with the receptor binding thereby cause the cleavage of HA₀, followed by acidic pH-mediated fusogenic rearrangement of HA2 and thereby prevent the viral entry related factors. Since 60% similarities were found in the estimated free energy and common binding interaction with 3 amino acids between Oseltamivir and 1-monocaproin, 1-monocaproin is also expected to

exhibit similar inhibiting effect against H_5N_1 hemagglutinin. Hence based on the molecular interaction study and common binding interaction of the compounds, the isolated compound 1-monocaproin will also behave like Oseltamivir. The previous study with homology modeling and docking with 3D structure of the S-Adenosylmethionine synthase a key enzyme in *E. Maximum* with 1-monocaproin revealed that the synthesized 1-monocaproin was found to possess anti parasite effect as well (Maheswari and Revathi, 2016).

Conclusion

At present, the food industries employ MAGs merely as emulsifying agents or stabilizers and it is the right time that MAGs are considered most significantly as antimicrobial agents. Since no single MAGs exhibited a broad spectrum antimicrobial effect, it can be concluded that a combination of MAGs synthesized with C:06:0 to C:12:0 carbon containing fatty acids may be considered as general-purpose antimicrobials and preservatives as they exhibit their inhibitory activity at very low concentration itself and hence cost effective as well. As 1-monocaproin docking with H_5N_1 Hemagglutinin of avian influenza virus showed Ki value of 98.61µm is lesser than standard, 1-monocaproin could be a potent antiviral agent against AV H_5N_1 and perhaps

could provide the similar effect as that of Oseltamivir at higher dose. Hence the compounds synthesized for this study; 1-monocaproin and 1-monocaprylin remains promising for numerous applications and need to be explored further.

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Author's Contributions

Palaniappan Maheswari: Designed the research plan, participated in all experiments, coordinated the data-analysis and writing the manuscript.

Kasthuri Revathi: Contributed in the experimental design, organised for the study and reviewed the manuscript.

Ethics

The authors declare that they have no competing interests in this work.

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